6-HYDROXYRAMULOSIN - A NEW METABOLITE FROM PESTALOTIA RAMULOSA

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During an investigation of ramulosin production (1) by Pestalotia ramulosa several additional metabolites have been isolated. These include mellein (2) (R_f 0.55; $\left[\alpha\right]_{D}^{25}$ -95.4° (<u>c</u> 1, MeOH)), two ene-polyyne derivatives $(R_{f}^{0.10} \text{ and } 0.15$, respectively), and only from shake cultures, a new metabolite (I). The latter (R $_{\rm f}$ 0.02; FeCl $_{\rm 3}$ spray, deep violet; yield, 30 mg/l) appeared 4-5 days after inoculation of a malt extract-peptone-dextrose (3) medium. Chloroform extracts of agitated cultures maintained at 16° for 8 days were examined by tlc (Mallinkrodt AR sheets, benzene development) which revealed the presence of these ultraviolet-absorbing components, including the major metabolite, ramulosin (R_f 0.38). After solvent evaporation, the residue was crystallized from hexane, and further purified by chromatography on Florisil $(CHCl_3$ eluent). Recrystallization from $CHCl_3$ -hexane provided (I), colorless needles, mp 132-133°, $\left[\alpha\right]_{D}^{25}$ + 91.6° (<u>c</u> 1, MeOH), $\lambda_{\max}^{\text{EtOH}}$ 262 m μ (\$10,530), KBr 3509 and 1070 cm⁻¹ (nonbonded secondary -OH). Calcd for $c_{10}H_{14}O_4$: C, 60.60; H, 7.07; mol wt 198. Found: C, 60.74; H, 7.20; mol wt 206 by vp osmometry; m/e 198. Its uv maximum was shifted to 291 m μ with alkali and was restored by acidification, a change similar to the behavior of ramulosin.

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Since I was comparable to ramulosin in many of its physical and chemical properties and contained but one more oxygen atom, placement of this function as a secondary hydroxyl group at the 4, 5, 6, or 7 position was indicated by the ir data. On biogenetic grounds the representation of I as 6-hydroxy-ramulosin is suggested by the occurrence of several 8-hydroxy-6-methoxy-and 6,8-dihydroxyisocoumarins(4). An attempt to confirm this structure by chemical conversion of I to the known 6,8-dihydroxy-3-methylisocoumarin(4a,5) by dehydrogenation with Pd/C in boiling p-cymene, led instead to a mixture of (-)-mellein (2) and 8-hydroxy-3-methylisocoumarin (6).

However, the 100 MHz nmr spectrum of I in comparison to that of ramulosin (Figure 1, b and a, respectively) provided proof for assignment of the additional oxygen at position 6, and gave an insight into the stereochemistry of the molecule. Thus, the H₃ multiplet at δ 4.46 (d of d of q, lH, J ae 2 Hz, J aa 12 Hz, J CHCH₃ 7 Hz) indicates that a -CH₂CHCH₃ group is present, with the methyl group pseudoequatorial. The triplet ($_{\rm Vic}$ 12 Hz) pattern of the H band at δ 2.85 indicates that the proton is transdiaxial to two other protons and thus excludes the possibility of an axial -OH at position 5. The presence of -OH at position 7 has been eliminated on the grounds that three allylic protons appear in the δ 2-3 region. A narrow band due to the carbinol methine proton at δ 4.29 (t of t, lH, $_{\rm Vic}$ 2 and 4 Hz) indicates that the proton is equatorial, and therefore, the -OH group has the 6-axial conformation. This is further confirmed since the C₇ methylene protons at δ 2.39 and 2.60 ($_{\rm Jgem}$ 19Hz) show no $_{\rm Jaa}$ splitting.

The fact that I provided levorotatory mellein upon chemical dehydrogenation accords with the ancillary finding that (-)-mellein was isolated from the \underline{P} .

ramulosa fermentation. This, together with the assignment of the 3-methyl of I as pseudoequatorial (Figure Ib), strongly suggests the $3\underline{R}$ absolute configuration for I, for the levorotatory of the two known (7) mellein enantiomers, and for

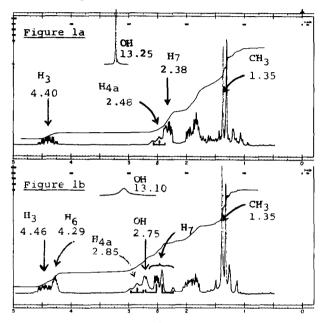


Figure I. 100 MHz Nmr Spectra of Ramulosin (a) and of I (b) in \mathtt{CDCl}_3

ramulosin. This agrees with earlier work which has shown the $3\underline{R}$ configuration for levorotatory 5-methylmellein (8) and for the 5-chloromellein-7-carboxylic acid derived from ochratoxin (9), by their degradations to D- β -hydroxybutyric acid. The metabolic branchpoints at which I and ramulosin, obviously biogenetically related not only to the isocoumarins but also to \underline{C} -acetyl- \underline{O} -orsellinic acid (4a,5), become transformed into hydroaromatic ring systems, are under further investigation.

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